

Poly[[bis(4,4'-bipyridine)bis(μ -oxalato)-trizinc(II)]-di- μ -4,4'-bipyridine- μ -oxalato]Karen J. Nordell,^{a*} Khadine A. Higgins^a and Mark D. Smith^b^aDepartment of Chemistry, Lawrence University, Appleton, Wisconsin 54912, USA, and^bDepartment of Chemistry and Biochemistry, University of South Carolina, Columbia, South Carolina 29208, USACorrespondence e-mail:
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Key indicators

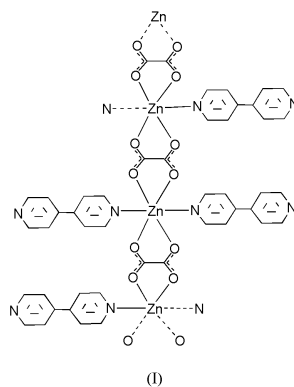
Single-crystal X-ray study
T = 293 K
Mean σ (C–C) = 0.003 Å
R factor = 0.027
wR factor = 0.062
Data-to-parameter ratio = 13.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The hydrothermally prepared title compound, $[\text{Zn}_3(\text{C}_{10}\text{H}_8\text{N}_2)_4(\text{C}_2\text{O}_4)_3]$, isostructural with the Fe analog, consists of undulating chains of oxalate-bridged Zn centers linked into infinite two-dimensional layers by bridging 4,4'-bipyridine ligands. An additional 4,4'-bipyridine ligand is coordinated to two of the three inequivalent Zn centers as a terminal ligand.

Received 31 January 2003
Accepted 13 February 2003
Online 21 February 2003

Comment

The title compound, (I), is isostructural with a previously reported iron analog (Zheng *et al.*, 1999). Three crystallographically inequivalent octahedral Zn^{2+} centers are linked into undulating chains along [010] by three inequivalent bridging oxalate groups. The chain subunits are further linked into infinite two-dimensional layers by two of the four crystallographically independent 4,4'-bipyridine ligands. Two additional 4,4'-bipyridine ligands coordinate to Zn1 and Zn3 in a terminal mode (Fig. 1), creating a 'self-interdigitating' layered network. The layers stack along [101].



The Zn–O distances [range 2.060 (1)–2.091 (1) Å; Table 1] and Zn–N distances [range 2.150 (2)–2.337 (2) Å; Table 1] are consistent with the reported Fe–O distances, and in good agreement with those in the related zinc–oxalato–4,4'-bipyridine layered polymer $[\text{Zn}(\text{ox})(\text{bipy})]$ (Lu *et al.*, 1999).

Experimental

The title compound was prepared by hydrothermal reaction of $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ (0.576 g, 0.2 mmol) with 4,4'-bipyridine (0.0160 g, 0.1 mmol) and oxalic acid (0.0124 g, 0.1 mmol) in water (0.80 ml) in an evacuated sealed Pyrex tube. The reaction was heated to 418 K at 10 K min^{-1} and held at that temperature for 48 h before cooling slowly (0.2 K min^{-1}) to 298 K. The reaction yielded a plentiful supply of yellow crystals.

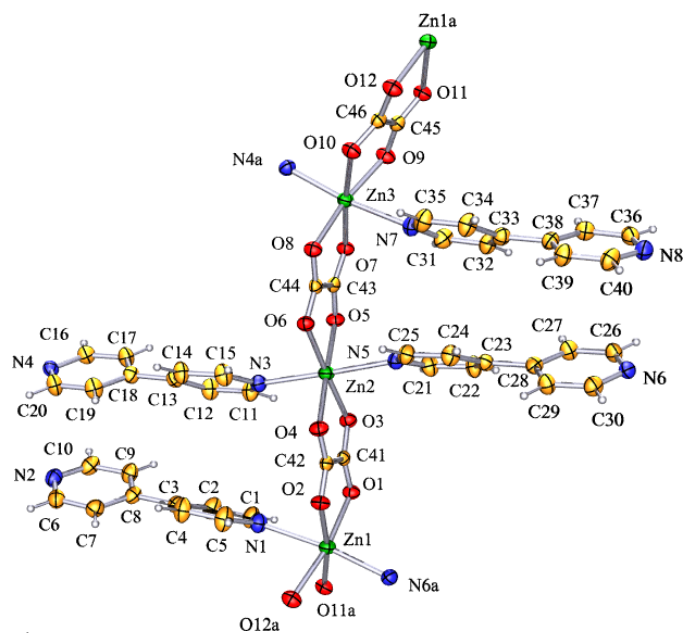


Figure 1
View of (I), with the atom-labeling scheme. Displacement ellipsoids are drawn at the 50% probability level. Atoms with suffix 'a' are symmetry generated; see Table 1 for details.

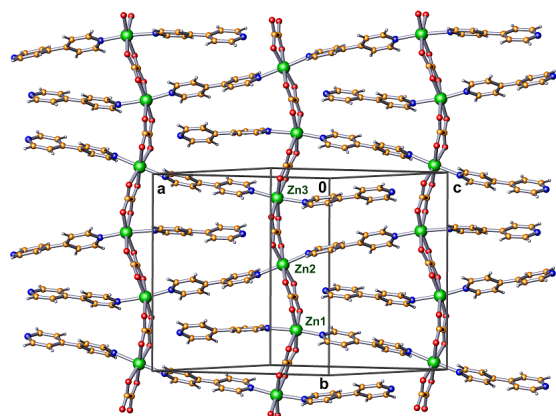


Figure 2
A single layer of (I), showing the undulating chains of oxalato-bridged Zn atoms and terminal and bridging 4,4'-bipyridine ligands.

Crystal data

$[\text{Zn}_3(\text{C}_{10}\text{H}_8\text{N}_2)_4(\text{C}_2\text{O}_4)_3]$
 $M_r = 1084.91$
 Monoclinic, $P2_1/c$
 $a = 16.2513(9) \text{ \AA}$
 $b = 15.5891(9) \text{ \AA}$
 $c = 16.376(1) \text{ \AA}$
 $\beta = 94.000(1)^\circ$
 $V = 4138.6(4) \text{ \AA}^3$
 $Z = 4$

$D_x = 1.741 \text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation
 Cell parameters from 9955 reflections
 $\theta = 2.6\text{--}26.4^\circ$
 $\mu = 1.81 \text{ mm}^{-1}$
 $T = 293(2) \text{ K}$
 Bar, yellow
 $0.42 \times 0.40 \times 0.22 \text{ mm}$

Data collection

Bruker SMART APEX CCD area-detector diffractometer
 ω scans
 Absorption correction: multi-scan (SADABS; Bruker, 1999)
 $T_{\min} = 0.415$, $T_{\max} = 0.594$
 32100 measured reflections

8443 independent reflections
 6454 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$
 $\theta_{\max} = 26.4^\circ$
 $h = -14 \rightarrow 20$
 $k = -18 \rightarrow 19$
 $l = -20 \rightarrow 20$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.027$
 $wR(F^2) = 0.062$
 $S = 1.03$
 8443 reflections
 622 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0256P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.002$
 $\Delta\rho_{\max} = 0.47 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.37 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (\AA).

Zn1—O2	2.0630 (13)	Zn2—O5	2.0880 (14)
Zn1—O1	2.0668 (13)	Zn2—N3	2.2152 (15)
Zn1—O12 ⁱ	2.0749 (14)	Zn2—N5	2.2419 (15)
Zn1—O11 ⁱ	2.0806 (14)	Zn3—O7	2.0601 (13)
Zn1—N6 ⁱⁱ	2.1601 (15)	Zn3—O8	2.0800 (13)
Zn1—N1	2.3370 (16)	Zn3—O10	2.0886 (14)
Zn2—O3	2.0733 (13)	Zn3—O9	2.0910 (14)
Zn2—O6	2.0782 (13)	Zn3—N4 ⁱⁱⁱ	2.1497 (15)
Zn2—O4	2.0866 (14)	Zn3—N7	2.2669 (16)

Symmetry codes: (i) $x, y - 1, z$; (ii) $-x, y - \frac{1}{2}, \frac{3}{2} - z$; (iii) $1 - x, \frac{1}{2} + y, \frac{1}{2} - z$.

H atoms were geometrically idealized, with $\text{C—H} = 0.93 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{iso}}(\text{C})$.

Data collection: SMART-NT (Bruker, 1999); cell refinement: SAINT-Plus-NT (Bruker, 1999); data reduction: SAINT-Plus-NT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

Funding was provided by Lawrence University and by the National Science Foundation through grant DMR:9873570.

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